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Chun-Sheng Zheng, Nan Yang, Mei Li and Zuo-Liang Jing*

College of Sciences, Tianjin University of Science and Technology, Tianjin 300222, People's Republic of China

Correspondence e-mail: jzl74@tust.edu.cn

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.028 wR factor = 0.070 Data-to-parameter ratio = 13.4

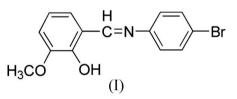
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-[(4-Bromophenyl)iminomethyl]-6-methoxyphenol

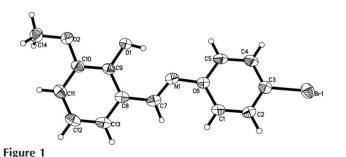
The title compound, $C_{14}H_{12}BrNO_2$, was prepared from 2-hydroxy-3-methoxybenzaldehyde and 4-bromophenylamine. The 4-hydroxy-3-methoxyphenyl and bromobenzene groups are *trans* with respect to the imino bond. The planarity of the molecule is supported by the conjugation of the imino group and the aromatic system, together with a resonanceassisted intramolecular hydrogen bond $[O-H \cdots N =$ 2.610 (3) Å]. Received 8 September 2005 Accepted 30 September 2005 Online 12 October 2005

Comment

The synthesis and structure determination of the title compound, (I) (Fig. 1), have been carried out in the context of the design of novel crystalline solids based on crystal engineering (Tynan *et al.*, 2005; Parashar *et al.*, 1988). The bond lengths of C7–C8 [1.429 (4) Å] and C7=N1 [1.277 (4) Å] are in good agreement with the values observed in a similar system (Jing *et al.*, 2005). The 4-bromophenylimine moiety (C1–C6/N1/Br1) is planar, with an r.m.s. deviation of fitted atoms of 0.0046 Å. The *o*-vanillin moiety (C7–C13/O1/O2) is planar, with an r.m.s. deviation of fitted atoms of 0.0055 Å. The dihedral angle between these two planes is 12.74 (8)°. The values of the N1–C7–C8–C13 and N1–C6–C5–C4 torsion angles of –178.9 (3) and 179.9 (3)°, respectively, illustrate the planarity of the molecule.



An intramolecular resonance-assisted hydrogen bond (Gigli *et al.*, 1989) (Table 2) contributes to the planarity of the molecule. Crystal packing is *via* $C-H(ar)\cdots Br$ interactions [3.704 (4) Å] connecting molecules along the *c* axis, and $C-H\cdots O$ interactions [3.346 (3) Å] involving an Osp^3 atom (Lommerse *et al.*, 1997; Molčanov *et al.*, 2004) (Table 2).



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Experimental

ethanol solution of 2-hydroxy-3-methoxy-An anhydrous benzaldehyde (1.52 g, 10 mmol) was added to an anhydrous ethanol solution of 4-bromophenylamine (1.72 g, 10 mmol), and the mixture was stirred at 350 K for 5 h under nitrogen. A yellow precipitate was obtained. The product was isolated, recrystallized from ethanol and dried in vacuo to give pure compound (I) in 83% yield. Single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution of (I).

Crystal data

C ₁₄ H ₁₂ BrNO ₂ $M_r = 306.16$ Orthorhombic, $P2_12_12_1$ a = 4.9175 (9) Å b = 12.524 (2) Å c = 20.551 (4) Å V = 1265.6 (4) Å ³ Z = 4 $D_x = 1.607$ Mg m ⁻³ Data collection	Mo $K\alpha$ radiation Cell parameters from 3003 reflections $\theta = 2.6-24.6^{\circ}$ $\mu = 3.24 \text{ mm}^{-1}$ T = 293 (2) K Prism, colourless $0.26 \times 0.24 \times 0.12 \text{ mm}$
Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 1999) $T_{\min} = 0.420, T_{\max} = 0.678$ 6772 measured reflections	2214 independent reflections 1993 reflections with $I > 2\sigma(I)$ $R_{int} = 0.019$ $\theta_{max} = 25.0^{\circ}$ $h = -5 \rightarrow 5$ $k = -14 \rightarrow 12$ $l = -23 \rightarrow 24$
Refinement Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.028$	$w = 1/[\sigma^2(F_o^2) + (0.0257P)^2 + 0.2106P]$

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0257P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.028$	+ 0.2106P]
$wR(F^2) = 0.070$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.07	$(\Delta/\sigma)_{\rm max} < 0.001$
2214 reflections	$\Delta \rho_{\rm max} = 0.38 \ {\rm e} \ {\rm \AA}^{-3}$
165 parameters	$\Delta \rho_{\rm min} = -0.41 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Absolute structure: Flack (1983),
	with 883 Friedel pairs

Table 1

Selected geometric parameters (Å, °).

Br1-C3	1.899 (3)	N1-C7	1.277 (4)
O2-C10	1.357 (3)	N1-C6	1.406 (4)
O2-C14	1.413 (4)		
C10-O2-C14	117.5 (2)	C1-C6-N1	125.1 (3)
C7-N1-C6	121.7 (2)	N1-C7-C8	123.3 (3)
C5-C6-N1	117.5 (3)		

Flack parameter: 0.004 (12)

Table 2

Hydrogen-bond	geometry	(A,	°).
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$D - \mathbf{H} \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O1-H1···N1	0.82	1.88	2.610 (3)	147
$C2-H2\cdot\cdot\cdot O2^i$	0.93	2.50	3.346 (3)	152
$C11-H11\cdots Br1^{ii}$	0.93	2.88	3.704 (4)	149

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + \frac{1}{2}, -y + 1, z + \frac{1}{2}$.

The H atom of the hydroxyl group was found in a difference; it and the other H atoms were included and refined with free coordinates and an isotropic U parameter. The other H atoms were included in calculated positions and refined using a riding-model approximation, with C-H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic CH, C-H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl CH₃, and O-H = 0.82 Å and $U_{iso}(H) = 1.2U_{eq}(O)$ for the hydroxyl group.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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References

- Bruker (1999). SADABS, SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Gigli, G., Belluci, F., Ferretti, V. & Bertolasi, V. (1989). J. Am. Chem. Soc. 111, 1023-1028.
- Jing, Z.-L., Yu, M., Chen, X., Diao, C.-H., Deng, Q.-L. & Fan, Z. (2005). Acta Cryst. E61, 0145-0146.

Lommerse, J. P., Price, S. L. & Taylor, R. (1997). J. Comput. Chem. 18, 757-774.

Molčanov, K., Kojić-Prodić, B. & Raos, N. (2004). Acta Cryst. B60, 424-432. Parashar, R. K., Sharma, R. C., Kumar, A. & Mohan, G. (1988). Inorg. Chim.

Acta, 151, 201-208.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Sheldrick G. M. (1997b). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.

Tynan, E., Jensen, P., Lees, A. C., Moubaraki, B., Murray, K. S. & Kruger, P. E. (2005). Cryst. Eng. Comm. 7, 90-95.